# i-STAT CG8+ Cartridge

Intended for use with the i-STAT 1 Analyzer (REF 04P75-01 & 03P75-06)

# NAME

i-STAT CG8+ Cartridge - REF 03P88-25

# **INTENDED USE**



The i-STAT CG8+ cartridge with the i-STAT 1 System is intended for use in the *in vitro* quantification of sodium, potassium, ionized Calcium, glucose, hematocrit, pH, oxygen partial pressure, and carbon dioxide partial pressure in arterial, venous or capillary whole blood.

Analyte	Intended Use							
Sodium (Na)	Sodium measurements are used for monitoring electrolyte imbalances.							
Potassium (K)	Potassium measurements are used in the diagnosis and monitoring of diseases and clinical conditions that manifest high and low potassium levels.							
Ionized Calcium (iCa)	lonized calcium measurements are used in the diagnosis, monitoring, and treatment of conditions including, but not limited to, parathyroid disease, a variety of bone diseases, chronic renal disease, tetany, and disturbances related to surgical and intensive care.							
Glucose (Glu)	Glucose measurements are used in the diagnosis, monitoring, and treatment of carbohydrate metabolism disorders including, but not limited to, diabetes mellitus, neonatal hypoglycemia, idiopathic hypoglycemia, and pancreatic islet cell carcinoma.							
Hematocrit (Hct)	Hematocrit measurements can aid in the determination and monitoring of normal or abnormal total red cell volume status including, but not limited to, conditions such as anemia, erythrocytosis, and blood loss related to trauma and surgery.							
рН	pH, $PO_2$ , and $PCO_2$ measurements are used in the diagnosis,							
Oxygen Partial Pressure ( <b>P</b> O <sub>2</sub> )	<ul> <li>monitoring, and treatment of respiratory disturbances and metabolic and respiratory-based acid-base disturbances.</li> </ul>							
Carbon Dioxide Partial Pressure ( <b>P</b> CO <sub>2</sub> )	Bicarbonate is used in the diagnosis and treatment of numerous potentially serious disorders associated with changes in body acid-base balance.							

# SUMMARY AND EXPLANATION / CLINICAL SIGNIFICANCE

### Measured:

### Sodium (Na)

Tests for sodium in the blood are important in the diagnosis and treatment of patients suffering from hypertension, renal failure or impairment, cardiac distress, disorientation, dehydration, nausea and diarrhea. Some causes of increased values for sodium include dehydration, diabetes insipidus, salt poisoning, skin losses, hyperaldosteronism and CNS disorders. Some causes for decreased values for sodium include dilutional hyponatremia (cirrhosis), depletional hyponatremia and syndrome of inappropriate ADH.

### Potassium (K)

Tests for potassium in the blood are important in the diagnosis and treatment of patients suffering from hypertension, renal failure or impairment, cardiac distress, disorientation, dehydration, nausea and diarrhea. Some causes of increased values for potassium include renal glomerular disease, adrenocortical insufficiency, diabetic ketoacidosis (DKA), sepsis and *in vitro* hemolysis. Some causes of decreased values for potassium include renal tubular disease, hyperaldosteronism, treatment of DKA, hyperinsulinism, metabolic alkalosis and diuretic therapy.

### Ionized Calcium (iCa)

Although most of the calcium in blood is bound to protein or complexed to smaller anionic species, the biologically active fraction of calcium is free ionized calcium. Through its role in a number of enzymatic reactions and in membrane transport mechanisms, ionized calcium is vitally important in blood coagulation, nerve conduction, neuromuscular transmission and in muscle contraction. Increased ionized calcium (hypercalcemia) may result in coma. Other symptoms reflect neuromuscular disturbances, such as hyperreflexia and/or neurologic abnormalities such as neurasthenia, depression or psychosis. Decreased ionized calcium (hypocalcemia) often results in cramps (tetany), reduced cardiac stroke work and depressed left ventricular function. Prolonged hypocalcemia may result in bone demineralization (osteoporosis) which can lead to spontaneous fractures. Measurements of ionized calcium have proven of value under the following clinical conditions: transfusion of citrated blood, liver transplantation, open heart surgery, neonatal hypocalcemia, renal disease, hyperparathyroidism, malignancy, hypertension and pancreatitis.

### Glucose (Glu)

Glucose is a primary energy source for the body and the only source of nutrients for brain tissue. Measurements for determination of blood glucose levels are important in the diagnosis and treatment of patients suffering from diabetes and hypoglycemia. Some causes for increased values of glucose include diabetes mellitus, pancreatitis, endocrine disorders (e.g., Cushing's syndrome), drugs (e.g., steroids, thyrotoxicosis), chronic renal failure, stress, or I.V. glucose infusion. Some causes of decreased values of glucose include insulinoma, adrenocortical insufficiency, hypopituitarism, massive liver disease, ethanol ingestion, reactive hypoglycemia, and glycogen storage disease.

### Hematocrit (Hct)

Hematocrit is a measurement of the fractional volume of red blood cells. This is a key indicator of the body's state of hydration, anemia or severe blood loss, as well as the blood's ability to transport oxygen. A decreased hematocrit can be due to either overhydration, which increases the plasma volume, or a decrease in the number of red blood cells caused by anemias or blood loss. An increased hematocrit can be due to loss of fluids, such as in dehydration, diuretic therapy, and burns, or an increase in red blood cells, such as in cardiovascular and renal disorders, polycythemia vera, and impaired ventilation.

### рΗ

pH is an index of the acidity or alkalinity of the blood with an arterial pH of <7.35 indicating an acidemia and >7.45 alkalemia  $^{1}$ .

### Oxygen Partial Pressure (PO2)

 $PO_2$  (partial pressure of oxygen) is a measurement of the tension or pressure of oxygen dissolved in blood. Some causes for decreased values of  $PO_2$  include decreased pulmonary ventilation (e.g., airway obstruction or trauma to the brain), impaired gas exchange between alveolar air and pulmonary capillary blood (e.g., bronchitis, emphysema, or pulmonary edema), and alteration in the flow of blood within the heart or lungs (e.g., congenital defects in the heart or shunting of venous blood into the arterial system without oxygenation in the lungs).

### Carbon Dioxide Partial Pressure (PCO<sub>2</sub>)

 $PCO_2$  along with pH is used to assess acid-base balance.  $PCO_2$  (partial pressure of carbon dioxide), the respiratory component of acid-base balance, is a measure of the tension or pressure of carbon dioxide dissolved in the blood.  $PCO_2$  represents the balance between cellular production of  $CO_2$  and ventilatory removal of  $CO_2$  and a change in  $PCO_2$  indicates an alteration in this balance. Causes of primary respiratory distress syndrome, and chronic obstructive pulmonary disease. Causes of primary respiratory alkalosis (decreased  $PCO_2$ ) are hypoxia (resulting in hyperventilation) due to chronic heart failure, edema and neurologic disorders, and mechanical hyperventilation.

## **TEST PRINCIPLE**

The i-STAT System uses direct (undiluted) electrochemical methods. Values obtained by direct methods may differ from those obtained by indirect (diluted) methods.<sup>2</sup>

### **Measured:**

### Sodium (Na), Potassium (K) and ionized Calcium (iCa)

The respective analyte is measured by ion-selective electrode potentiometry. In the calculation of results, concentration is related to potential through the Nernst equation.

### Glucose (Glu)

Glucose is measured amperometrically. Oxidation of glucose, catalyzed by the enzyme glucose oxidase, produces hydrogen peroxide ( $H_2O_2$ ). The liberated  $H_2O_2$  is oxidized at the electrode to produce a current proportional to the sample glucose concentration.

$$\beta$$
-D-glucose + H<sub>2</sub>O + O<sub>2</sub>   
 $H_2O_2$    
 $H_2O_2$    
 $H_2O_2$    
 $D$ -gluconic acid + H<sub>2</sub>O<sub>2</sub>  
 $D$ -gluconic acid + H<sub>2</sub>O<sub>2</sub>

### Hematocrit (Hct)

Hematocrit is determined conductometrically. The measured conductivity, after correction for electrolyte concentration, is inversely related to the hematocrit.

pН

pH is measured by direct potentiometry. In the calculation of results for pH, concentration is related to potential through the Nernst equation.

### **PO**<sub>2</sub>

 $PO_2$  is measured amperometrically. The oxygen sensor is similar to a conventional Clark electrode. Oxygen permeates through a gas permeable membrane from the blood sample into an internal electrolyte solution where it is reduced at the cathode. The oxygen reduction current is proportional to the dissolved oxygen concentration.

### $PCO_2$

 $PCO_2$  is measured by direct potentiometry. In the calculation of results for  $PCO_2$ , concentration is related to potential through the Nernst equation.

### **Temperature "Correction" Algorithm**

pH,  $\mathbf{P}O_2$ , and  $\mathbf{P}CO_2$  are temperature-dependent quantities and are measured at 37°C. The pH,  $\mathbf{P}O_2$ , and  $\mathbf{P}CO_2$  readings at a body temperature other than 37°C can be 'corrected' by entering the patient's temperature on the chart page of the analyzer. In this case, blood gas results will be displayed at both 37°C and the patient's temperature.

The pH,  $PO_2$ , and  $PCO_2$  at the patient's temperature (T<sub>p</sub>) are calculated as follows <sup>3</sup>:

$$pH(T_p) = pH - 0.0147(T_p - 37) + 0.0065(7.4 - pH)(T_p - 37)$$

$$PO_{2}(T_{p}) = PO_{2} \times 10^{\frac{5.49 \times 10^{-11} PO_{2}^{3.88} + 0.071}{9.72 \times 10^{-9} PO_{2}^{3.88} + 2.30}} (T_{p} - 37)$$

$$PCO_2(T_p) = PCO_2 \times 10^{0.019(T_p - 37)}$$

### **Calculated:**

### HCO<sub>3</sub>, TCO<sub>2</sub>, and BE

- HCO<sub>3</sub> (bicarbonate), the most abundant buffer in the blood plasma, is an indicator of the buffering capacity of blood. Regulated primarily by the kidneys, HCO<sub>3</sub> is the metabolic component of acid-base balance.
- TCO<sub>2</sub> is a measure of carbon dioxide which exists in several states: CO<sub>2</sub> in physical solution or loosely bound to proteins, bicarbonate (HCO<sub>3</sub>) or carbonate (CO<sub>3</sub>) anions, and carbonic acid (H<sub>2</sub>CO<sub>3</sub>). Measurement of TCO<sub>2</sub> as part of an electrolyte profile is useful chiefly to evaluate HCO<sub>3</sub> concentration. TCO<sub>2</sub> and HCO<sub>3</sub> are useful in the assessment of acid-base imbalance (along with pH and *P*CO<sub>2</sub>) and electrolyte imbalance.
- The calculated TCO<sub>2</sub> provided by the i-STAT System is determined from the measured and reported values of pH and *P*CO<sub>2</sub> according to a simplified and standardized form of the Henderson-Hasselbalch equation.<sup>3</sup>
- This calculated TCO<sub>2</sub> measurement is metrologically traceable to the i-STAT pH and PCO<sub>2</sub> measurements, which are in turn traceable to primary standard reference materials for pH and PCO<sub>2</sub>. Like all calculated parameters reported by the i-STAT System, the user can independently determine TCO<sub>2</sub> values from the reported pH and PCO<sub>2</sub> measurements using a combination of the equation for HCO<sub>3</sub> given in the PCO<sub>2</sub>.
- Base excess of the extracellular fluid (ECF) or standard base excess is defined as the concentration of titratable base minus the concentration of titratable acid when titrating the average ECF (plasma plus interstitial fluid) to an arterial plasma pH of 7.40 at *P*CO<sub>2</sub> of 40 mmHg at 37 °C. Excess concentration of base in the average ECF remains virtually constant during acute changes in the *P*CO<sub>2</sub> and reflects only the non-respiratory component of pH-disturbances.

When a cartridge includes sensors for both pH and  $PCO_2$ , bicarbonate (HCO<sub>3</sub>), total carbon dioxide (TCO<sub>2</sub>) and base excess (BE) are calculated.<sup>3</sup>

 $\begin{array}{l} \log HCO_3 = pH + \log \textit{P}CO_2 - 7.608 \\ TCO_2 = HCO_3 + 0.03\textit{P}CO_2 \\ BE_{ecf} = HCO_3 - 24.8 + 16.2(pH-7.4) \\ BE_b = (1 - 0.014^*Hb) * [ HCO_3 - 24.8 + (1.43^*Hb + 7.7) * (pH - 7.4) ] \end{array}$ 

sO<sub>2</sub>

- sO<sub>2</sub> (oxygen saturation) is the amount of oxyhemoglobin expressed as a fraction of the total amount of hemoglobin able to bind oxygen (oxyhemoglobin plus deoxyhemoglobin).
- sO<sub>2</sub> is calculated from measured *P*O<sub>2</sub> and pH and from HCO<sub>3</sub> calculated from measured *P*CO<sub>2</sub> and pH. However, this calculation assumes normal affinity of oxygen for hemoglobin. It does not take into account erythrocyte diphosphoglycerate (2,3-DPG) concentrations which affect the oxygen dissociation curve. The calculation also does not take into account the effects of fetal hemoglobin or dysfunctional hemoglobins (carboxy-, met-, and sulfhemoglobin). Clinically significant errors can result from incorporation of such an estimated sO<sub>2</sub> value for oxygen saturation in further calculations, such as shunt fraction, or by assuming the value obtained is equivalent to fractional oxyhemoglobin.

$$sO_2 = 100 = \frac{(X^3 + 150X)}{X^3 + 150X + 23400}$$
  
where X =  $PO_2 \cdot 10^{(0.48(pH-7.4)-0.0013(HCO_3-25))}$ 

### Hemoglobin

The i-STAT System provides a calculated hemoglobin result which is determined as follows 4:

hemoglobin (g/dL) = hematocrit (% PCV) x 0.34

hemoglobin (g/dL) = hematocrit (decimal fraction) x 34

To convert a hemoglobin result from g/dL to mmol/L, multiply the displayed result by 0.621. The calculation of hemoglobin from hematocrit assumes a normal MCHC.

See below for information on factors affecting results. Certain substances, such as drugs, may affect analyte levels in vivo. <sup>5</sup> If results appear inconsistent with the clinical assessment, the patient sample should be retested using another cartridge.

# REAGENTS

### Contents

Each i-STAT cartridge contains one reference electrode, sensors for the measurement of specific analytes, and a buffered aqueous calibrant solution that contains known concentrations of analytes and preservatives. A list of reactive ingredients relevant for the i-STAT CG8+ cartridge is indicated below:

Sensor	Reactive Ingredient	<b>Biological Source</b>	Minimum Quantity
Na	Sodium (Na⁺)	N/A	121 mmol/L
К	Potassium (K <sup>+</sup> )	N/A	3.6 mmol/L
iCa	Calcium (Ca <sup>2+</sup> )	N/A	0.9 mmol/L
Glu	Glucose	N/A	7 mmol/L
Giù	Glucose Oxidase	Aspergillus niger	0.002 IU
pH Hydrogen Ion (H <sup>+</sup> )		N/A	6.66 pH
<b>P</b> CO <sub>2</sub>	PCO <sub>2</sub> Carbon Dioxide (CO <sub>2</sub> )		25.2 mmHg

### Warnings and Precautions

- For in vitro diagnostic use.
- Cartridges are intended for single-use only. Do not reuse.
- Refer to the i-STAT 1 System Manual for all warnings and precautions.

### Storage Conditions

- Refrigerated at 2-8°C (35-46°F) until expiration date.
- Room Temperature at 18-30°C (64-86°F). Refer to the cartridge box for shelf life.

### **INSTRUMENTS**

The i-STAT CG8+ cartridge is intended for use with the i-STAT 1 analyzer REF 04P75-01 (Model 300-G) and REF 03P75-06 (Model 300W).

# SPECIMEN COLLECTION AND PREPARATION FOR ANALYSIS

### Specimen Types

Arterial, venous or capillary whole blood. Sample Volume: 95 µL

### Blood Collection Options and Test Timing (time from collection to cartridge fill)

Analyte	Syringes	Test Timing	Evacuated Tubes	Test Timing	Capillary Tubes	Test Timing
lonized Calcium pH	Without anticoagulant	3 minutes	Without anticoagulant	3 minutes	With balanced heparin anticoagulant	3 minutes
PCO <sub>2</sub> PO <sub>2</sub>	<ul> <li>With balanced heparin anticoagulant (or lithium heparin anticoagulant for pH, PCO<sub>2</sub>, and PO<sub>2</sub> only) (syringe must be filled per manufacturer's recommendation)</li> <li>Maintain anaerobic conditions.</li> <li>Remix thoroughly before filling cartridge.</li> </ul>	10 minutes	<ul> <li>With lithium heparin anticoagulant (tubes must be filled per manufacturer's recommendation)</li> <li>Maintain anaerobic conditions.</li> <li>Remix thoroughly before filling cartridge.</li> </ul>	10 minutes	or lithium heparin if labeled for the measurement of electrolytes (for pH, <i>P</i> CO <sub>2</sub> , and <i>P</i> O <sub>2</sub> only)	
Cadium	Without anticoagulant	3 minutes	Without anticoagulant	3 minutes	With balanced heparin anticoagulant	3 minutes
Sodium Potassium Glucose Hematocrit	ium heparin e anticoagulant or		With lithium heparin anticoagulant (tubes must be filled per manufacturer's recommendation)	30 minutes or lithium heparin if labeled for the measurement of electrolytes		

Analyte	Syringes	Test Timing	Evacuated Tubes	Test Timing	Capillary Tubes	Test Timing
	<ul> <li>manufacturer's</li> <li>recommendation)</li> <li>Remix thoroughly before filling cartridge.</li> </ul>		<ul> <li>Remix thoroughly before filling cartridge,</li> </ul>			

### PROCEDURE FOR CARTRIDGE TESTING

Each cartridge is sealed in a foil pouch for protection during storage--do not use if pouch has been punctured.

- A cartridge should not be removed from its protective pouch until it is at room temperature (18-30 °C or 64-86 °F). For best results, the cartridge and analyzer should be at room temperature.
- Since condensation on a cold cartridge may prevent proper contact with the analyzer, allow refrigerated cartridges to equilibrate at room temperature for 5 minutes for a single cartridge and 1 hour for an entire box before use.
- Use a cartridge immediately after removing it from its protective pouch. Prolonged exposure may cause a cartridge to fail a Quality Check.
- Do not return unopened, previously refrigerated cartridges to the refrigerator.
- o Cartridges may be stored at room temperature for the time frame indicated on the cartridge box.

Filling and Sealing the Cartridge (after cartridge has been equilibrated and blood sample has been collected)

- 1. Place the cartridge on a flat surface.
- 2. Mix the sample thoroughly. Invert a lithium heparin blood collection tube at least 10 times. If sample was collected into a syringe, invert syringe for 5 seconds then roll the syringe between the palms (hands parallel to the ground) for 5 seconds, flip and roll for an additional 5 seconds. The blood in the hub of the syringe will not mix, therefore expelling 2 drops before filling a cartridge is desired. Note that it may be difficult to properly mix a sample in a 1.0 mL syringe.
- 3. Fill the cartridge immediately after mixing. Direct the hub of syringe or tip of the transfer device (capillary tube, pipette, or dispensing tip) into the sample well of the cartridge.
- 4. Slowly dispense sample into the sample well until the sample reaches the fill mark indicated on the cartridge. Cartridge is properly filled when the sample reaches the 'fill to' mark and a small amount of sample is in the sample well. The sample should be continuous, no bubbles or breaks (see System Manual for details).
- 5. Fold the snap closure over the sample well.

### **Performing Patient Analysis**

- 1. Press the power button to turn on the handheld.
- 2. Press 2 for *i-STAT Cartridge*.
- 3. Follow the handheld prompts.
- 4. Scan the lot number on the cartridge pouch.
- 5. Continue normal procedures for preparing the sample, and filling and sealing the cartridge.
- 6. Push the sealed cartridge into the handheld port until it clicks into place. Wait for the test to complete.
- 7. Review the results.

For additional information for cartridge testing, refer to the i-STAT 1 System Manual located at www.pointofcare.abbott.

### Analysis Time

Approximately 130–200 seconds

### **Quality Control**

The i-STAT quality control regimen comprises four aspects, with a system design that reduces the opportunity for error, including:

- 1. A series of automated, on-line quality measurements that monitors the sensors, fluidics, and instrumentation each time a test is performed.
- 2. A series of automated, on-line procedural checks that monitors the user each time a test is performed.
- 3. Liquid materials are available to be used to verify the performance of a batch of cartridges when they are first received or when storage conditions are in question. The performance of this procedure is not a manufacturer's system instruction.
- 4. Traditional quality control measurements that verify the instrumentation using an independent device, which simulates the characteristics of the electrochemical sensors in a way that stresses the performance characteristics of the instrumentation.

For additional information on Quality Control, refer to the i-STAT 1 System Manual located at www.pointofcare.abbott.

### **Calibration Verification**

Calibration Verification is a procedure intended to verify the accuracy of results over the entire measurement range of a test. The performance of this procedure is not a manufacturer's system instruction. However, it may be required by regulatory or accreditation bodies. While the Calibration Verification Set contains five levels, verification of the measurement range could be accomplished using the lowest, highest and mid-levels.

## **EXPECTED VALUES**

		REPORTABLE	REFERENC	E RANGE
TEST	UNITS *	RANGE	arterial	venous
MEASURED				
Sodium/Na	mmol/L(mEq/L)	100-180	138-1	
Potassium/K	mmol/L(mEq/L)	2.0-9.0	3.5–4.9	96**
iCa	mmol/L	0.25-2.50	1.12–1	.32 7
10a	mg/dL	1.0-10.0	4.5–5	.3 7
	mmol/L	1.1–38.9	3.9–5	.8 7
Glu	mg/dL	20-700	70–10	)5 <sup>7</sup>
	g/L	0.20-7.00	0.70–1	.05 7
	% PCV ***	15–75	38–51	
Hematocrit/Hct	Fraction	0.15–0.75	0.38–0	.51 6
рН		6.50 - 8.20	7.35 - 7.45 <sup>7</sup>	7.31 - 7.41*****
<b>P</b> O <sub>2</sub>	mmHg	5 - 800	80 - 105 <sup>6</sup> *****	
	kPa	0.7 – 106.6	10.7 - 14.0 <sup>6</sup> *****	
<b>P</b> CO <sub>2</sub>	mmHg	5 - 130	35 - 45 <sup>7</sup>	41 - 51
	kPa	0.67 – 17.33	4.67 - 6.00	5.47 - 6.80
CALCULATED				
	g/dL	5.1–25.5	12–17	
Hemoglobin/Hb	g/L	51–255	120–1	
	mmol/L	3.2–15.8	7–11	1 6
Bicarbonate/	mmol/L	1.0 – 85.0	22 – 26*****	23 – 28*****
HCO₃	(mEq/L)	1.0 00.0	22 20	20 20
TCO <sub>2</sub>	mmol/L	5 - 50	23 - 27	24 - 29
	(mEq/L)	0 00	20 2.	2. 20
Base Excess/ BE	mmol/L	(-30) - (+30)	(-2) – (+3) <sup>7</sup>	(-2) – (+3) <sup>7</sup>
	(mEq/L)			
sO₂	%	0-100	95 –	90

\* The i-STAT System can be configured with the preferred units. Not applicable for pH test.

 The reference range for potassium has been reduced by 0.2 mmol/L from the range cited in Reference <sup>6</sup> to account for the difference in results between serum and plasma results.

\*\*\* PCV, packed cell volume.

\*\*\*\* The reference ranges for hematocrit and hemoglobin span both female and male populations

\*\*\*\*\* The reference ranges shown are for a healthy population. Interpretation of blood gas measurements depend on the underlying condition (e.g., patient temperature, ventilation, posture and circulatory status).

\*\*\*\*\*\* Calculated from Siggard-Andersen nomogram.<sup>1</sup>

### Unit Conversion:

- **Ionized Calcium (iCa):** To convert mmol/L to mg/dL, multiply the mmol/L value by 4. To convert mmol/L to mEq/L, multiply the mmol/L value by 2.
- Glucose (Glu): To convert mg/dL to mmol/L, multiply the mg/dL value by 0.055.
- Hematocrit (Hct): To convert a result from % PCV (packed cell volume) to fraction packed cell volume, divide the % PCV result by 100. For the measurement of hematocrit, the i-STAT System can be customized to agree with methods calibrated by the microhematocrit reference method using either K<sub>3</sub>EDTA or K<sub>2</sub>EDTA anticoagulant. Mean cell volumes of K<sub>3</sub>EDTA anticoagulated blood are approximately 2–4% less than K<sub>2</sub>EDTA anticoagulated blood. While the choice of anticoagulant affects the microhematocrit method to which all hematocrit methods are calibrated, results from routine samples on hematology analyzers are independent of the anticoagulant used. Since most clinical hematology analyzers are calibrated by the microhematocrit method using K<sub>3</sub>EDTA anticoagulant, the i-STAT System default customization is K<sub>3</sub>EDTA.
- **PO<sub>2</sub> and PCO<sub>2</sub>:** To convert **P**O<sub>2</sub> and **P**CO<sub>2</sub> results from mmHg to kPa, multiple the mmHg value by

### 0.133.

The reference ranges programmed into the analyzer and shown above are intended to be used as guides for the interpretation of results. Since reference ranges may vary with demographic factors such as age, gender and heritage, it is recommended that reference ranges be determined for the population being tested.

# METROLOGICAL TRACEABILITY

The measured analytes in the i-STAT CG8+ cartridge are traceable to the following reference materials or methods. The i-STAT System controls and calibration verification materials are validated for use only with the i-STAT System and assigned values may not be commutable with other methods.

### Sodium (Na) and Potassium (K) and Ionized Calcium (iCa)

The respective analyte values assigned to i-STAT System controls and calibration verification materials are traceable to the U.S. National Institute of Standards and Technology (NIST) standard reference material SRM956.

### Glucose (Glu)

The i-STAT System test for glucose measures glucose amount-of-substance concentration in the plasma fraction of arterial, venous, or capillary whole blood (dimension mmol L<sup>-1</sup>) for *in vitro* diagnostic use. Glucose values assigned to i-STAT System Controls and calibration verification materials are traceable to the U.S. National Institute of Standards and Technology (NIST) standard reference material SRM965.

### Hematocrit (Hct)

The i-STAT System test for hematocrit measures packed red blood cell volume fraction in arterial, venous, or capillary whole blood (expressed as the % packed cell volume) for *in vitro* diagnostic use. Hematocrit values assigned to i-STAT working calibrators are traceable to the Clinical and Laboratory Standards Institute (CLSI) H7-A3 procedure for determining packed cell volume by the microhematocrit method.<sup>8</sup>

## рΗ

The i-STAT System test for pH measures the hydrogen ion amount-of-substance concentration in the plasma fraction of arterial, venous, or capillary whole blood (expressed as the negative logarithm of the relative molal hydrogen ion activity) for *in vitro* diagnostic use. pH values assigned to i-STAT System controls and calibration verification materials are traceable to the U.S. National Institute of Standards and Technology (NIST) standard reference materials SRMs 186-I, 186-II, 185, and 187.

# The i-STAT System test for oxygen partial pressure measures oxygen partial pressure in arterial, venous, or capillary whole blood (dimension kPa) for *in vitro* diagnostic use. **P**O<sub>2</sub> values assigned to i-STAT System controls and calibration verification materials are traceable to U.S. National Institute of Standards and Technology (NIST) standard reference materials via commercially available certified specialty medical gas standards.

### $PCO_2$

The i-STAT System test for carbon dioxide partial pressure measures carbon dioxide partial pressure in arterial, venous, or capillary whole blood (dimension kPa) for *in vitro* diagnostic use. **P**CO<sub>2</sub> values assigned to i-STAT System controls and calibration verification materials are traceable to U.S. National Institute of Standards and Technology (NIST) standard reference materials via commercially available certified specialty medical gas standards.

Additional information regarding metrological traceability is available from Abbott Point of Care Inc.

### PERFORMANCE CHARACTERISTICS

The typical performance data summarized below was collected in health care facilities by health care professionals trained in the use of the i-STAT System and comparative methods.

### Precision

Precision data was collected in multiple sites as follows: Duplicates of each control fluid were tested in the morning and in the afternoon on five days for a total of 20 replicates. The averaged statistics are presented below.

Test	Units	Aqueous Control	Mean	SD (Standard Deviation)	CV (%) [Coefficient of Variation (%)]
Na	mmol/L or mEq/L	Level 1	120.0	0.46	0.4
		Level 3	160.0	0.53	0.3
K	mmol/L or mEq/L	Level 1	2.85	0.038	1.3
		Level 3	6.30	0.039	0.6
iCa	mmol/L	Level 1	1.60	0.017	1.1
		Level 3	0.84	0.012	1.4
Glu	mg/dL	Level 1	41.8	0.68	1.6
	-	Level 3	289	2.4	0.8
Hct	% PCV	Low	30.0	0.44	1.5
	(packed cell	High	49.0	0.50	1.0
	volume)				
pН		Level 1	7.165	0.005	0.08
		Level 3	7.656	0.003	0.04
<b>P</b> O <sub>2</sub>	mmHg	Level 1	65.1	3.12	4.79
	-	Level 3	146.5	6.00	4.10
<b>P</b> CO <sub>2</sub>	mmHg	Level 1	63.8	1.57	2.5
	-	Level 3	19.6	0.40	2.0

### Method Comparison

Method comparison data was collected using CLSI guideline EP9-A.9

Deming regression analysis <sup>10</sup> was performed on the first replicate of each sample. In the method comparison table, n is the number of specimens in the data set,  $S_{xx}$  and  $S_{yy}$  refer to estimates of imprecision based on the duplicates of the comparative and the i-STAT methods respectively,  $S_{y.x}$  is the standard error of the estimate, and r is the correlation coefficient.\*

Method comparisons will vary from site to site due to differences in sample handling, comparative method calibration and other site-specific variables.

\* The usual warning relating to the use of regression analysis is summarized here as a reminder. For any analyte, "if the data is collected over a narrow range, the estimate of the regression parameters are relatively imprecise and may be biased. Therefore, predictions made from these estimates may be invalid". <sup>10</sup> The correlation coefficient, r, can be used as a guide to assess the adequacy of the comparative method range in overcoming this problem. As a guide, the range of data can be considered adequate if r>0.975.

Sodium/Na (mmol/L or mEq/L)		Beckman Synchron CX®3	Kodak Ektachem <sup>™</sup> 700	Nova STAT Profile <sup>®</sup> 5
Venous blood samples were	n	189	142	192
collected in lithium heparin	S <sub>xx</sub>	0.74	0.52	0.54
Vacutainer <sup>®</sup> tubes and analyzed in duplicate on the i-STAT System.	Syy	0.53	0.58	0.53
A portion of the specimen was	Slope	1.00	0.98	0.95
centrifuged and the separated	Inťt	-0.11	3.57	5.26
plasma was analyzed in duplicate on comparative methods within 20 minutes of collection.	S <sub>y.x</sub>	1.17	1.04	1.53
	X <sub>min</sub>	126	120	124
	X <sub>max</sub>	148	148	148

	r	0.865	0.937	,		0.838	
		Beckman					
Potassium/K (mmol/L or mEq/L)		Synchron CX®3	Koda	ak Ektad	chem™ 700	Nova STAT Profile <sup>®</sup> 5	
Venous blood samples were	n	189	142			192	
collected in lithium heparin	Sxx	0.060	0.031			0.065	
Vacutainer <sup>®</sup> tubes and analyzed in duplicate on the i-STAT System.	Syy	0.055	0.059	)		0.055	
A portion of the specimen was	Slope	0.97	1.06			0.99	
centrifuged and the separated	Int't	0.02	-0.15			-0.01	
plasma was analyzed in duplicate on	Sy.x	0.076	0.060	)		0.112	
comparative methods within 20 minutes of collection.	Xmin	2.8	3.0			2.8	
minutes of collection.	Xmax	5.7	9.2			5.8	
	r	0.978	0.993			0.948	
lonized Calcium/iCa (mmol/L)		Radiomet ICA1	er			STAT	
Venous blood samples were	n	47				57	
collected in lithium heparin	Sxx	0.009			0.	017	
Vacutainer <sup>®</sup> tubes and analyzed in	Syy	0.017			0.	017	
duplicate on the i-STAT System and on the comparative methods within	Slope	0.925			0.	960	
10 minutes of each other.	Int't	0.113			0.	0.062	
	Sy.x	0.035			0.029		
	Xmin	0.46			0.53		
	Xmax	2.05			.05		
	r	0.982 0			982		
01		Deekmen				Dade	
Glucose/Glu (mg/dL)		Beckman Coulter LX2	0®	Ba	yer 860	Dimension RxL- Xpand	
Venous blood samples were	n	35			40	32	
collected in lithium heparin	Sxx	2.21		4.71		0.98	
Vacutainer <sup>®</sup> tubes and analyzed in duplicate on the i-STAT System.	Syy	0.69		0.96		0.59	
duplicate of the Porter System.	Slope	1.03		0.99		1.01	
A portion of the specimen was	Int't	-3.39		-1.67		-0.85	
centrifuged and the separated plasma was analyzed in duplicate on	Sy.x	0.91			0.70	1.57	
comparative methods within 20	Xmin	45		58		48	
minutes of collection.	Xmax	297		167		257	
	r	0.999		(	).993	0.998	
Hematocrit/Hct (% PCV) (% packed cell volume)		Coulter <sup>®</sup> S Plus		ova AT ile <sup>®</sup> 5	Abbott Cell-Dyn 4000	Sysmex SE9500	
Venous blood samples, collected in	n	142	192		29	29	
lithium heparin Vacutainer <sup>®</sup> tubes,	Sxx	0.50	0.46		0.41	0.53	
were analyzed in duplicate on the	Syy	1.09	1.31		0.77	0.76	
i-STAT System and on the comparative methods for hematocrit	Slope	0.98	1.06		1.06	1.11	
within 20 minutes of collection.	Int't	1.78	-3.98		-1.42	-4.19	
	Sy.x	2.03	2.063		1.13	0.98	
	Xmin	18	21		19	24	
	Xmax	51	50		46	47	
		0.952	0.932		0.993	0.980	

рН		IL BGE	Radiometer ICA 1	S	ova IAT file 5	Radiometer ABL500
Venous blood samples were collected in	n	62	47	57		45
evacuated tubes and arterial samples	Sxx	0.005	0.011	0.00	)6	0.004
were collected in blood gas syringes with lithium heparin anticoagulant.	Syy	0.009	0.008	0.00	)8	0.008
All samples were analyzed in duplicate	Slope	0.974	1.065	1.05	58	1.0265
on the i-STAT System and on the	Int't	0.196	-0.492	-0.4	36	-0.1857
comparative methods within 10 minutes of each other. Arterial blood samples	Sy.x	0.012	0.008	0.01	0	0.0136
were collected from hospital patients in 3	Xmin	7.210	7.050	7.05	50	
mL blood gas syringes and were	Xmax	7.530	7.570	7.57	'0	
analyzed in duplicate on the i-STAT System and the comparative method within 5 minutes of each other.	r	0.985	0.990	0.99		0.986
Oxygen Partial Pressure/ <i>P</i> O <sub>2</sub> (mmHg)		Radiometer ABL500	Radiome ABL70			Bayer 845
Arterial blood samples were collected	n	45	29	30		
from hospital patients in 3 cc blood gas	Sxx	3.70	2.04	3.03		
syringes and were analyzed in duplicate on the i-STAT System and the	Syy	2.78	2.64 3.28			
comparative method within 5 minutes of	Slope	1.023	0.962 1.033		3	
each other.	Int't	-2.6	1.2		-2.9	
	Sy.x	2.52	3.53		3.44	
	Xmin		39		31	
	Xmax		163		185	
	r	0.996	0.990		0.996	6
Carbon Dioxide Partial Pressure/PCO <sub>2</sub> (mmHg)		IL B	GE		Radior	neter ABL500
Venous blood samples were collected	n	62		29		
in blood gas syringes.	Sxx	0.69		0.74		
All samples were analyzed in duplicate	Syy	1.24		0.53		
on the i-STAT System and on the comparative methods within 10 minutes	Slope	1.003		1.016	;	
of each other. Arterial blood samples	Int't	-0.8		1.1		
were collected from hospital patients in	Sy.x	1.65		0.32		
3 cc blood gas syringes and were	Xmin	30.4		28		
analyzed in duplicate on the	Xmax	99.0	91			

### 99.0 Xmax i-STAT System and the comparative method within 5 minutes of each other. 0.989 r

# FACTORS AFFECTING RESULTS

The following substances were evaluated in plasma for relevant analytes at the test concentrations recommended in CLSI guideline EP7-A2 <sup>11</sup> unless otherwise noted. For those identified as an interferant the interference is described.

Substance	Test Concentration (mmol/L)	Analyte	Interference (Yes/No)	Comment
Acetaldehyde	0.045 <sup>12</sup>	Glu	No	
Acctominophon	taminophen 1.32	Na	No	
Acetaminophen	1.32	К	No	

0.999

Substance	Test Concentration (mmol/L)	Analyte	Interference (Yes/No)	Comment
		iCa	Yes	Decreased results
		Glu	No	
Acetaminophen	0.132 12	iCa	No	
(therapeutic)	0.152	Glu	No	
		Na	No	
Acetylcysteine	10.2	K	No	
Accipicysteme	10.2	iCa	Yes	Decreased results
		Glu	Yes	Decreased results
Acetylcysteine	0.3 <sup>13 14</sup>	iCa	No	
(therapeutic)		Glu	No	
Acetoacetate	2.0	Glu	No	
		Na	No	
Ascorbate	0.34	K	No	
ASCOLDAIC	0.04	iCa	No	
		Glu	No	
		Na	Yes	Increased results. Use another method.
		к	Yes	Increased results and rate of star (***) outs. Use another method.
Bromide	37.5	iCa	Yes	Increased results. Use another method.
		Glu	Yes	Decreased results. Use another method.
		Hct	Yes	Increased rate of star (***) outs
		Na	No	· · ·
Duo unido		K	No	
Bromide	2.5 <sup>15 16 17</sup>	iCa	No	
(therapeutic)		Glu	Yes	Decreased results
		Hct	No	
		Na	No	
0. Hudrow but roto	6.0 <sup>18</sup>	K	No	
β-Hydroxybutyrate	0.0	iCa	No	
		Glu	No	
Dopamine	0.006	Glu	No	
Formaldehyde	0.133 <sup>12</sup>	Glu	No	
Hydroxyurea	0.92	Glu	Yes	Increased results. Use another method.
		Na	No	
		К	No	
Lactate	6.6	iCa	Yes	Decreased results by up to 0.07 mmol/L.
		Glu	No	
Leflunomide	0.03	iCa	Yes	Decreased results
		Na	No	
Magnesium Chloride		K	No	
	1.0	iCa	Yes	Increased results by up to 0.04 mmol/L.
Maltose	13.3	Glu	No	
		Na	Yes	Increased results
	1			
Nithiodote (sodium thiosulfate)	16.7 <sup>19</sup>	K	Yes	Decreased results

Substance	Test Concentration (mmol/L)	Analyte	Interference (Yes/No)	Comment
		Glu	Yes	Decreased results
Pyruvate	0.31	Glu	No	
		Na	No	
Salicylate	4.34	К	No	
Salicylate	4.34	iCa	Yes	Decreased results
		Glu	No	
Salicylate (therapeutic)	0.5 20	iCa	Yes	Decreased results by up to 0.03 mmol/L
Thiocyanate	6.9	iCa	Yes	Decreased results. Use another method
		Glu	Yes	Decreased results
Thiocyanate (therapeutic)	0.5 <sup>12</sup>	Glu	No	
Uric Acid	1.4	Glu	No	

The degree of interference at concentrations other than those reported above might not be predictable. It is possible that interfering substances other than those tested may be encountered.

- Relevant comments regarding interference of Acetaminophen, Acetylcysteine, Bromide, Hydroxyurea, Leflunomide, Nithiodote, and Salicylate are noted below:
  - Acetaminophen has been shown to interfere with i-STAT ionized calcium results at a concentration proscribed by the CLSI guideline, 1.32 mmol/L, which represents a toxic concentration. Acetaminophen at 0.132 mmol/L, which represents the upper end of the therapeutic concentration, has been shown not to significantly interfere with i-STAT ionized calcium results.
  - Acetylcysteine has been tested at two levels: the CLSI recommended level of 10.2 mmol/L and a concentration of 0.30 mmol/L. The latter is 3 times the peak therapeutic plasma concentration associated with treatment to reverse acetaminophen poisoning. APOC has not identified a therapeutic condition that would lead to levels consistent with the CLSI recommended level.
  - Bromide has been tested at two levels: the CLSI recommended level and a therapeutic plasma concentration level of 2.5 mmol/L. The latter is the peak plasma concentration associated with halothane anesthesia, in which bromide is released. APOC has not identified a therapeutic condition that would lead to levels consistent with the CLSI recommended level.
  - Hydroxyurea has been shown to interfere with glucose results at 0.92 mmol/L. Hydroxyurea is a DNA synthesis inhibitor used in the treatment of various forms of cancer, sickle cell anemia, and HIV infection. This drug is used to treat malignancies including melanoma, metastatic ovarian cancer, and chronic myelogenous leukemia. It is also used in the treatment of polycythemia vera, thrombocythemia, and psoriasis. At typical doses ranging from 500 mg to 2 g/day, concentrations of hydroxyurea in patients' blood may be sustained at approximately 100 to 500 µmol/L. Higher concentrations may be observed soon after dosing or at higher therapeutic doses.
  - Leflunomide has been shown to interfere with iCa results at 0.03 mmol/L. Leflunomide is an isoxazole immunomodulatory agent that inhibits dihydroorotate dehydrogenase, an enzyme involved in *de novo* pyrimidine synthesis, and that has antiproliferative activity. It is used in the treatment of some immune diseases. Following oral administration, leflunomide is metabolized to an active metabolite, teriflunomide, which is responsible for essentially all its *in vivo* activity. The active metabolite teriflunomide reaches a plasma concentration of 8.5 µg/mL (0.031 mmol/L) after a 100 mg loading dose and the steady state concentration is maintained at 63 µg/mL (0.23 mmol/L) after 24 weeks of maintenance dose at 25 mg/day <sup>21</sup> when treating inflammatory polyarthropathy.
  - Nithiodote (sodium thiosulfate) has been shown to interfere with sodium, potassium, ionized calcium, and glucose results at 16.7 mmol/L. Nithiodote (sodium thiosulfate) is indicated for the treatment of acute cyanide poisoning. The journal article titled "Falsely increased chloride and

missed anion gap elevation during treatment with sodium thiosulfate" indicated that sodium thiosulfate could be used in the treatment of calciphylaxis indicating that "the highest concentration likely to be seen in plasma [is] after infusion of a 12.5 g dose of sodium thiosulfate pentahydrate. Assuming that the 12.5 g dose of sodium thiosulfate pentahydrate is distributed in a typical blood volume of 5 L with a hematocrit of 40%, the peak sodium thiosulfate plasma concentration expected is 16.7 mmol/L." <sup>19</sup>

 Salicylate has been shown to significantly decrease ionized calcium results at a concentration proscribed by the CLSI guideline, 4.34 mmol/L, which represents a toxic concentration. Salicylate at 0.5 mmol/L, which represents the upper end of the therapeutic concentration, has been shown to decrease ionized calcium results by approximately 0.03 mmol/L.

Factor	Analyte	Effect
Sodium heparin	Na	Sodium heparin may increase sodium results up to 1 mmol/L. <sup>22</sup>
	INd	Venous stasis (prolonged tourniquet application) and forearm exercise
Venous stasis	iCa	may increase ionized calcium due to a decrease in pH caused by
	104	localized production of lactic acid. <sup>23</sup>
		Venous stasis (prolonged tourniquet application) and forearm exercise
	рН	may decrease pH due to localized production of lactic acid.
Line draw	Hct	Low hematocrit results can be caused by contamination of flush solutions
		in arterial or venous lines.
		Back flush a line with a sufficient amount of blood to remove intravenous
		solutions, heparin, or medications that may contaminate the sample.
		Five to six times the volume of the catheter, connectors, and needle is
		recommended.
		Heparin binds calcium. Each unit of heparin added per mL of blood will
		decrease ionized calcium by 0.01 mmol/L <sup>23</sup> Therefore, the correct ratio of heparin anticoagulant to blood must be achieved during sample
	iCa	collection. Intravenous injection of 10,000 units of heparin has been
Heparin		shown in adults to cause a significant decrease of ionized calcium of
		about 0.03 mmol/L. <sup>23</sup> Use only non-heparinized sample transfer
		devices when using the i-STAT system aqueous control and calibration
		verification materials.
	iCa	Exposing the sample to air will cause an increase in pH due to loss of
	ica	CO <sub>2</sub> , which will decrease ionized calcium.
		Exposure of the sample to air will cause an increase in $PO_2$ when values
Exposing the	<b>P</b> O <sub>2</sub>	are below 150 mmHg and a decrease in $PO_2$ when values are above
sample to air		150 mmHg (approximate <b>P</b> O <sub>2</sub> of room air).
	pH POO	
	<b>P</b> CO <sub>2</sub>	Exposing the sample to air allows $CO_2$ to escape which causes $PCO_2$ to
	HCO <sub>3</sub> TCO <sub>2</sub>	decrease and pH to increase and $HCO_3$ and $TCO_2$ to be underestimated.
	1002	Hemodilution of the plasma by more than 20% associated with priming
	Na	cardiopulmonary bypass pumps, plasma volume expansion or other
		fluid administration therapies using certain solutions may cause clinically
	iCa	significant error on sodium, chloride, ionized calcium and pH results.
Hemodilution		These errors are associated with solutions that do not match the ionic
		characteristics of plasma. To minimize these errors when hemodiluting
	рН	by more than 20%, use physiologically balanced multi-electrolyte
		solutions containing low-mobility anions (e.g., gluconate).
Cold temperature	<b>P</b> O <sub>2</sub>	Do not ice samples before testing as <b>PO</b> <sub>2</sub> results may be falsely elevated
		in cold samples. Do not use a cold cartridge as <b>P</b> O <sub>2</sub> results may be
		falsely decreased if the cartridge is cold.
	ĸ	Potassium values will increase in iced specimens

### OTHER FACTORS AFFECTING RESULTS

Factor	Analyte	Effect
Allowing blood to stand (without exposure to air)	K	If heparinized whole blood is allowed to stand before testing, potassium values will first decrease slightly, then increase over time.
	Glu	Glucose values will decrease in whole blood samples over time. Venous blood glucose is as much as 7 mg/dL less than capillary blood glucose as a result of tissue utilization. <sup>24</sup>
	рН	pH decreases on standing anaerobically at room temperature at a rate of 0.03 pH units per hour. <sup>1</sup>
	<b>P</b> O <sub>2</sub>	Standing anaerobically at room temperature will decrease $PO_2$ at a rate of 2–6 mmHg per hour. <sup>1</sup>
	<b>P</b> CO <sub>2</sub>	Allowing blood to stand (without exposure to air) before testing will increase $PCO_2$ by approximately 4 mmHg per hour.
	HCO <sub>3</sub>	Calculated HCO <sub>3</sub> and TCO <sub>2</sub> results are over-estimated, if blood is
	TCO <sub>2</sub>	allowed to stand (without exposure to air), due to metabolic processes.
Sample type	к	Serum Potassium results may be 0.1 to 0.7 mmol/L higher than Potassium results from anticoagulated samples due to the release of Potassium from platelets <sup>2</sup> and red blood cells during the clotting process.
Sample mixing	Hct	Samples from 1 mL syringes should not be used to determine hematocrit if testing is delayed.
Hemolysis	к	Potassium values obtained from skin puncture samples may vary due to hemolysis or an increase in tissue fluid from improper technique during the collection procedure.
Under fill or partial draw	<b>P</b> CO <sub>2</sub>	The use of partial draw tubes (evacuated tubes that are adjusted to draw less than the tube volume, e.g., a 5 mL tube with enough vacuum to draw only 3 mL) is not recommended due to the potential for decreased
	HCO <sub>3</sub>	$PCO_2$ , HCO <sub>3</sub> and TCO <sub>2</sub> values. Underfilling blood collection tubes may also cause decreased $PCO_2$ , HCO <sub>3</sub> and TCO <sub>2</sub> results. Care must be
	TCO <sub>2</sub>	taken to eliminate "bubbling" of the sample with a pipette when filling a cartridge to avoid the loss of CO <sub>2</sub> in the blood.
pH dependence	Glu	The dependence of the i-STAT glucose test with respect to pH is as follows: values below pH 7.4 at 37 °C decrease results by approximately 0.9 mg/dL (0.05 mmol/L) per 0.1 pH unit. Values above pH 7.4 at 37 °C increase results by approximately 0.8 mg/dL (0.04 mmol/L) per 0.1 pH unit.
<b>P</b> O <sub>2</sub> dependence	Glu	The dependence of the i-STAT glucose test with respect to $PO_2$ is as follows: oxygen levels of less than 20 mmHg (2.66 kPa) at 37 °C may decrease results.
Method of calculation	sO <sub>2</sub>	Calculated $sO_2$ values from a measured $PO_2$ and an assumed oxyhemoglobin dissociation curve may differ significantly from the direct measurement. <sup>3</sup>
Clinical conditions	HCO₃	Causes of primary metabolic acidosis (decrease calculated HCO <sub>3</sub> ) are ketoacidosis, lactate acidosis (hypoxia), and diarrhea. Causes of primary metabolic alkalosis (increase calculated HCO <sub>3</sub> ) are vomiting and antacid treatment.
Erythrocyte sedimentation rate	Hct	<ul> <li>The measurement of certain blood samples with high erythrocyte sedimentation rates (ESR) may be affected by analyzer angle. While testing blood samples, beginning 90 seconds after the cartridge is inserted, the analyzer should remain level until a result is obtained. A level surface includes running the handheld in the downloader/ recharger.</li> <li>Hematocrit results can be affected by the settling of red blood cells in the collection device. The best way to avoid the effect of settling is to test the sample immediately. If there is a delay in testing of a minute or more, the sample must be remixed thoroughly.</li> </ul>

Factor	Analyte	Effect
White Blood Cell Count (WBC)	Hct	Grossly elevated white blood cell counts may increase results.
Lipids	Hct	Abnormally high lipids may increase results. Interference from lipids will be about two thirds the size of the interference from protein.
Total Protein	Hct	Hematocrit results are affected by the level of total protein as follows:         Displayed Result       Total Protein (TP) < 6.5 g/dL       Total Protein (TP) > 8.0 g/dL         HCT < 40% PCV
Sodium	Hct	The sample electrolyte concentration is used to correct the measured conductivity prior to reporting hematocrit results. Factors that affect sodium will therefore also affect hematocrit.
Propofol (Diprivan <sup>®</sup> ) or thiopental <u>sodium</u>	<b>P</b> CO <sub>2</sub>	The use of CG8+ cartridge is recommended, which is free from clinically significant interference at all relevant therapeutic doses.
₽O₂ sensitivity	<b>P</b> CO <sub>2</sub>	In patient samples where the $PO_2$ is > 100 mmHg above the normal range (80- 105 mmHg), an increase in $PCO_2$ of approximately 1.5 mmHg (with a range of 0.9 to 2.0 mmHg) may be observed for every 100 mmHg increase in $PO_2$ . For example, if an oxygenated patient has a measured $PO_2$ of 200 mmHg, and a normal $PO_2$ is 100 mmHg, the impact to the $PCO_2$ result may be increased by approximately 1.5 mmHg.

### **KEY TO SYMBOLS**

Symbol	Definition/Use
2	2 months room temperature storage at 18-30 <sup>o</sup> C
	Use by or expiration date. An expiration date expressed as YYYY-MM-DD means the last day the product can be used.
LOT	Manufacturer's lot number or batch code. The lot number or batch will appear adjacent to this symbol.
Σ	Sufficient for <n> tests</n>
EC REP	Authorized representative for Regulatory Affairs in the European Community.
X	Temperature limitations. The upper and lower limits for storage are adjacent to upper and lower arms.
REF	Catalog number, list number, or reference
$\otimes$	Do not reuse.
	Manufacturer
Ĩ	Consult instructions for use or see System Manual for instructions.
IVD	In vitro diagnostic medical device
CE	Compliance to the European directive on <i>in vitro</i> diagnostic devices (98/79/EC)
<b>Rx ONLY</b>	For prescription use only.

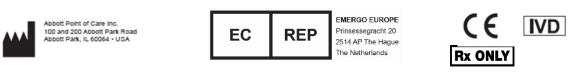
Additional Information: To obtain additional product information and technical support, refer to the company website at <u>www.pointofcare.abbott.</u>

# References

- 1. Pruden EL, Siggard-Andersen O, Tietz NW. Blood Gases and pH. In: C.A. Burtis and E.R. Ashwood, ed. *Tietz Textbook of Clinical Chemistry*. Second Edition ed. Philadelphia: W.B. Saunders Company; 1994.
- Tietz NW, Pruden EL, Siggaard-Andersen O. Electrolytes. In: C.A. Burtis and E.R. Ashwood, ed. *Tietz Textbook of Clinical Chemistry*. Second Edition ed. Philadelphia: W.B. Saunders Company; 1994.
- 3. CLSI. Blood Gas and pH Analysis and Related Measurements; Approved Guideline. *CLSI document C46-A*. 2001.
- 4. Evaluation of Formed Elements of Blood. In: Bower JD, Ackerman PG, Toto G, eds. *Clinical Laboratory Methods*. St. Louis: The C.V. Mosby Company; 1974.
- 5. Young DS. *Effects of Drugs on Clinical Laboratory Tests*. 3rd ed. ed. Washington, DC: American Association of Clinical Chemistry; 1990.
- 6. Statland BE. *Clinical Decision Levels for Lab Tests*. Oradell, NJ: Medical Economic Books; 1987.
- 7. Painter PC, Cope JY, Smith JL. Reference Ranges, Table 41–20. In: C.A. Burtis and E.R. Ashwood, ed. *Tietz Textbook of Clinical Chemistry*. Second Edition ed. Philadelphia: W.B. Saunders Company; 1994.
- 8. CLSI. Procedure for Determining Packed Cell Volume by the Microhematocrit Method; Approved Standard-Third Edition. *CLSI document H07-A3*. 2000.
- 9. CLSI. Method Comparison and Bias Estimation Using Patient Samples; Approved Guideline. *CLSI document EP9-A*. 1995.
- 10. Cornbleet PJ, Gochman N. Incorrect least-squares regression coefficients in method-comparison analysis. *Clinical Chemistry*. 1979;25(3).
- 11. Clinical and Laboratory Standards Institute. Interference Testing in Clinical Chemistry; Approved Guideline—Second Edition. *CLSI document EP7-A2*. 2005.
- 12. Wu AHB. Tietz Clinical Guide to Laboratory Tests: Elsevier Health Sciences; 2006.
- Whillier S, Raftos JE, Chapman B, Kuchel PW. Role of N-acetylcysteine and cystine in glutathione synthesis in human erythrocytes. *Redox Report*. 2009;14(3):115-121.
- 14. Ventura P, Panini R, Pasini MC, Scarpetta G, Salvioli G. N-acetyl-cysteine reduces homocysteine plasma levels after single intravenous administration by increasing thiols urinary excretion. *Pharmacological Research*. 1999;40(4):345-350.
- 15. Hankins DC, Kharasch ED. Determination of the halothane metabolites trifluoroacetic acid and bromide in plasma and urine by ion chromatography. *Journal of Chromatography B: Biomedical Applications*. May 1997;692(2):413-418.
- 16. Kharasch ED, Hankins D, Mautz D, Thummel KE. Identification of the enzyme responsible for oxidative halothane metabolism: Implications for prevention of halothane hepatitis. *Lancet*. May 1996;347(9012):1367-1371.
- 17. Morrison JE, Friesen RH. Elevated serum bromide concentrations following repeated halothane anaesthesia in a child. *Canadian Journal of Anaesthesia*. October 1990;37(7):801-803.

- Charles RA, Bee YM, Eng PHK, Goh SY. Point-of-care blood ketone testing: Screening for diabetic ketoacidosis at the emergency department. *Singapore Medical Journal*. November 2007;48(11):986-989.
- 19. Wendroth SM, Heady TN, Haverstick DM, et al. Falsely increased chloride and missed anion gap elevation during treatment with sodium thiosulfate. *Clinica Chimica Acta*. April 2014;431:77-79.
- Borthwick GM, Johnson AS, Partington M, Burn J, Wilson R, Arthur HM. Therapeutic levels of aspirin and salicylate directly inhibit a model of angiogenesis through a Cox-independent mechanism. *FASEB Journal*. October 2006;20(12):2009-2016.
- 21. Sanofi-Aventis Canada Inc. Product Monograph PrARAVA® Submission, Control No.: 187857. Date of Revision: December 23, 2015. Available at: <u>http://products.sanofi.ca/en/arava.pdf</u>.
- 22. Tips on Specimen Collection. In: Mark Zacharia, ed. *Vol 1. Monograph of Medical Laboratory Observer's "Tips from the Clinical Experts"*. Montvale NJ: Medical Economics in collaboration with Becton, Dickinson and Company; 1997.
- Fraser D, Jones G, Kooh SW, Raddle I. Calcium and Phosphate Metabolism. In: C.A. Burtis and E.R. Ashwood, ed. *Tietz Textbook of Clinical Chemistry*. Second Edition ed. Philadelphia: W.B. Saunders Company; 1994.
- 24. Young DS, Bermes EW. Influence of Site Collection on Blood Gases and pH. In: C.A. Burtis and E.R. Ashwood, ed. *Tietz Textbook of Clinical Chemistry*. Second Edition ed. Philadelphia: W.B. Saunders Company; 1994.
- 25. The Merck Index. Eleventh ed. NJ: Merck & Co., Inc.; 1989.

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